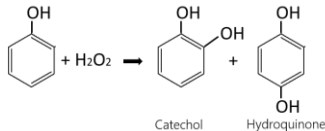


# ENGINEERING INTERCONNECTED-CHANNEL MONOLITHIC REACTORS: FROM COMSOL TO 3D PRINTING

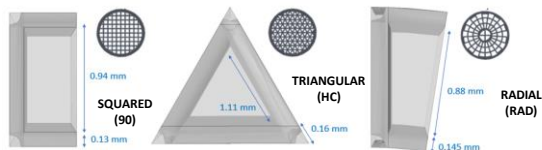
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## INTRODUCTION

- 3D Fe/SiC monoliths with interconnected channels printed by direct ink writing (Robocasting).
- 3D Fe/SiC monoliths are feasible catalytic reactors for phenol hydroxylation with hydrogen peroxide in water.

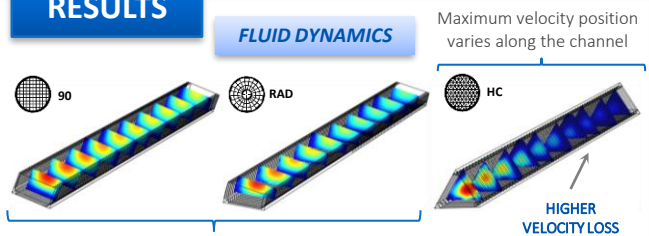


- CFD to study the effect of channel geometry on the oxidation performance (fluid dynamics, mass transport and kinetics).

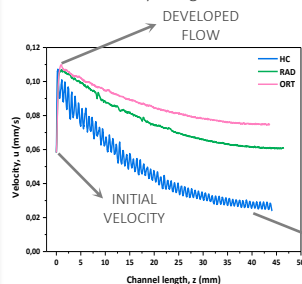


## RESULTS

### FLUID DYNAMICS



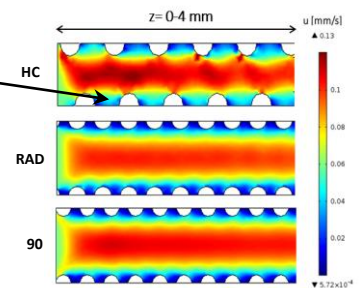
Maximum velocity along the centreline of the channel



- Oscillating movement in HC geometry caused by the angular contact of the interconnexions (which are not faced, as in the other geometries).
- Velocity loss in HC geometry increases residence time of the fluid in the interior of the channel

### OSCILLATING MOVEMENT DUE TO TURBULENCE

- Flow of the reactants through the interconnexions only in HC geometry

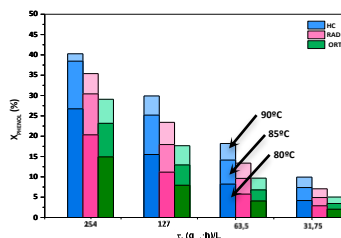


- HC geometry provides better contact between the catalyst and the reactants

- Pressure drop ( $\Delta P$ ): 90 < RAD < HC; but still negligible (max. 83 mPa)

### MASS TRANSPORT AND CHEMICAL REACTION

- External and internal mass transport are not limiting (data do not shown)  $\rightarrow$  Chemical reaction controls the overall rate



Triangular (HC) geometry provides superior performance

Exhibits the highest phenol conversion

## CONCLUSIONS

- Triangular shaped-channel monoliths provide the best performance for phenol hydroxylation. Therefore, triangular cells are selected as prototype to be robocasted in the 3D printer.
- Work is now on progress to experimentally validate the reactor modelling.

## REFERENCES

- [1] Vega G., et al 3D-PRINTING STRUCTURED CATALYSTS FOR THE SUSTAINABLE PRODUCTION OF DIHYDROBENZENES. ANQUE-ICC3 Conference. Santander (Spain), June 17-18, 2019.
- [2] A. Quintanilla et al. 3D-Printed Fe-doped silicon carbide monolithic catalysts for wet peroxide oxidation processes. Applied Catalysis B: Environmental, 2018. 235: p. 246-255.

## COMPUTATIONAL METHODS

COMSOL MULTIPHYSICS®

CFD MODULE

### 1. Mass transfer - Transport of diluted species

- Channel:** convection + external diffusion
- Porous wall:** effective internal diffusion + reaction

$$\nabla \cdot (-D_i \nabla c_i) + \mathbf{u} \cdot \nabla c_i = R_i$$

$$\mathbf{N}_i = -D_i \nabla c_i + \mathbf{u} c_i$$

$$\begin{cases} -(r_{H_2O_2}) = \frac{k_1 C_{H_2O_2}}{1 + k_2 C_{H_2O_2}} \left[ \frac{mol}{g_{cat} \cdot h} \right] \\ -(r_{phenol}) = \frac{k_{phenol} C_{phenol} C_{H_2O_2}}{1 + k_2 C_{H_2O_2}} \end{cases}$$

	$k_1$ (L/g <sub>cat</sub> ·h)	$k_2$ (L/mol)	$k_{phenol}$ (L <sup>2</sup> /mol·g <sub>cat</sub> ·h)
80°C	$1.66 \cdot 10^{-2}$	11.84	$2.31 \cdot 10^{-2}$
85°C	$1.54 \cdot 10^{-2}$	8.13	$3.05 \cdot 10^{-2}$
90°C	$1.50 \cdot 10^{-2}$	3.81	$2.85 \cdot 10^{-2}$

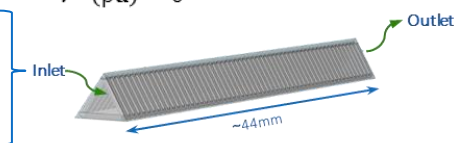
### 2. Fluid flow - Creeping flow interface (Re < 1)

Laminar flow with inertial term neglected

$$0 = \rho(\mathbf{u} \cdot \nabla) \mathbf{u} = \nabla \cdot [-p\mathbf{I} + \mu(\nabla \mathbf{u} + (\nabla \mathbf{u})^T)] + \mathbf{F}$$

$$\nabla \cdot (\rho \mathbf{u}) = 0$$

$C_{phenol,0} = 0.3M$   
 $C_{H_2O_2,0} = 0.3M$   
 $T = 80-90^\circ C$   
 $Q = 0.25-2 mL/min$   
 $W = 3.81 g$   
 $\tau = W/Q = 31.75-254 (g \cdot h)/L^{-1}$



Free Tetrahedral mesh with Normal element size

